

## SHORT REPORTS

### VANADIUM IN THE BROWN SEAWEED, *DESMARESTIA FIRMA*

BRIAN MAXSTEAD COCKERILL, PAUL FINCH and ELIZABETH PERCIVAL

Department of Chemistry, Bourne Laboratory, Royal Holloway College, Egham Hill, Egham, Surrey, TW20 0EX, U.K.

(Received 10 July 1978)

**Key Word Index**—*Desmarestia firma*; brown seaweed; alga; vanadium.

Both *Desmarestia firma* and *D. ligulata* contain free sulphuric acid (6% of the dry weight in the former species) [1]. No reason for this acid content has so far been discovered. Both species grow below low tide level. Ascidi-ans which grow submerged also contain free sulphuric acid (up to 1.0 molar), and it is considered [2] that this is connected with a high vanadium content (1300 ppm). It has been suggested [3] that the highly acidic environment results in the vanadium oscillating between the oxidation states of 3 and 4 using oxygen gas as the oxidant. To determine if vanadium played a similar role in *D. firma*, the seaweed was analysed for vanadium.

#### RESULTS AND DISCUSSION

The concentration of vanadium in the 5 samples varied from 1.50 to 2.50 ppm. Work by Yamamoto *et al.* [4] on species of Sargassaceae, Ishigeaceae and Spermatochnaceae found that the vanadium contents of these brown seaweeds varied from 0.82 to 10.5 ppm. Although *D. firma* does contain some vanadium the quantity is comparable with other seaweeds which are devoid of free sulphuric acid. It is therefore concluded that the sulphuric

acid in *D. firma* has a role different from that present in ascidians which possess an abnormally high content of vanadium, but could perhaps have the same function with another metal.

#### EXPERIMENTAL

The sample of *Desmarestia firma* was collected from 10 m depth in False Bay, Cape Town, South Africa, by Richard Simon and his colleagues in March, 1975, and was freeze-dried immediately after removal from seawater. Five different aliquots of weed (5 g each) were analysed for vanadium by the method of Jones and Watkinson [5].

#### REFERENCES

1. Carlberg, G. E., Percival, E. and Rhaman, O. (1978) *Phytochemistry* **17**, 1289.
2. Carlisle, D. P. (1968) *Proc. Roy. Soc.* **B171**, 31.
3. Gillard, R. D. personal communication.
4. Yamamoto, T., Fujita, T. and Ishibashi, M. (1970) *Rec. Oceanogr. Works Jpn* **10**, 127.
5. Jones, G. B. and Watkinson, J. H. (1959) *Analyt. Chem.* **31**, 1344.

### SESQUITERPENES FROM THE ESSENTIAL OIL OF *ASTRANTIA MAJOR*

H. BUURMA, R. BOS\*, D. H. E. TATTJE and J. H. ZWAVING

Laboratory of Pharmacognosy, State University, Groningen, Antonius Deusinglaan 2, The Netherlands

(Received 20 May 1978)

**Key Word Index**—*Astrantia major*; Umbelliferae; essential oil; sesquiterpenes;  $\beta$ -sinensal;  $\beta$ -sinensol;  $\beta$ -sinensyl acetate.

*Astrantia major* L., (family Umbelliferae, subfamily Saniculoideae) is a robust perennial found in Central Europe. The plant has previously been investigated for various constituents [1, 2]. However, although the essential oils of many species of the Umbelliferae have been extensively studied, very little is known about those

of *A. major* or other members of the subfamily Saniculoideae [2]. Since the chemical characters of these oils clearly have a high selective value for chemotaxonomic purposes [3] an investigation into the constituents of the essential oil of *Astrantia major* was started. This publication deals with the isolation and structural elucidation of three sesquiterpenes, isolated from the essential fruit oil.

\* Where reprints are available.